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AP<sup>®</sup> Chemistry 2008 Scoring Guidelines The College Board: Connecting Students to College Success The College Board is a not-for-profit membership association whose mission is to connect students to college success and ... Calculate the value of the standard free energy change, ...

## **AP Chemistry 2008 Scoring Guidelines - College Board**

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## **2008 AP CHEMISTRY FREE-RESPONSE QUESTIONS (Form**

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## **B)**

2008 AP<sup>®</sup> Chemistry Free-Response Questions The following comments on the 2008 free-response questions for AP<sup>®</sup> Chemistry were compiled by the Chief Reader, Eleanor Siebert of Mount St. Mary's College in Los Angeles, California. They give an overview of each free-response question and of how students performed on the question,

## **2008 Exam - AP Subjects**

AP Chemistry 2008 Free-Response Questions AP<sup>®</sup> Chemistry 2008 Free-Response Questions The College Board: ... Answer Questions 1, 2, and 3. The Section II score weighting for each question is 20 percent.  $C(s) + CO_2(g) \rightleftharpoons 2 CO(g)$  1.

## **2008 Ap Chemistry Free Response Answers Section 2**

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### **ap08 chemistry form b sgs - College Board**

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### **2008 AP Chemistry Free Response Questions? | Yahoo Answers**

AP\* Chemistry: 2008 Released Multiple Choice Exam NO CALCULATORS MAY BE USED Note: For all questions, assume that the temperature is 298 K, the pressure is 1.00 atmosphere, and solutions are aqueous unless otherwise specified.

Throughout the test the following symbols have the definitions

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specified unless otherwise noted. Part A

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## **AP Chemistry Exam Free-Response Question and Scoring**

...

AP\* Stoichiometry Free Response Questions KEY page 4 MM =  
 $(4.68 \times 0.60) / (0.5 \times 0.1) = 56$  or  $6 \times 10^1$  an alternate solution  
for (c) molality =  $0.5 \text{ }^\circ\text{C} / (4.68 \text{ } 0.5 \text{ }^\circ\text{C/m}) = 0.107 \text{ m mol solute}$   
 $= (0.107 \text{ mol} / \text{kg solvent}) \times 0.100 \text{ kg solvent} = 0.0107 \text{ mol}$

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## **AP\* Stoichiometry Free Response Questions KEY**

response must include specific information about both substances. (a) (b) (c) (d) At 250C and 1 atn, F2 is a gas, whereas 12 is a solid The melting point of NaF is 9930C, whereas the melting point of CsCl is 6450C. The shape of the ICIE ion is square planar, whereas the shape of the BF4- ion is tetrahedral.

## **AP Chemistry Intermolecular Forces FRQs**

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## **2008 Ap Chemistry Free Response Answers Form A**

AP\* General Equilibrium Free Response Questions page 2 1992.  
 $2 \text{NaHCO}_3(\text{s}) \rightarrow \text{Na}_2\text{CO}_3(\text{s}) + \text{H}_2\text{O}(\text{g}) + \text{CO}_2(\text{g})$ . Solid sodium hydrogen carbonate,  $\text{NaHCO}_3$ , decomposes on heating according to the equation above. (a) A sample of 100. grams of solid  $\text{NaHCO}_3$  was placed in a previously evacuated rigid 5.00-liter container and heated to  $160^\circ\text{C}$ .

## **AP\* General Equilibrium Free Response Questions**

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## **AP chemistry Free response 2008 Form B Question 3**

AP\* Solution Chemistry Free Response Questions page 2 1980  
(a) A solution containing 3.23 grams of an unknown compound dissolved in 100.0 grams of water freezes at  $-0.97^{\circ}\text{C}$ . The solution does not conduct electricity. Calculate the molecular weight of the compound. (The

## **AP\* Solution Chemistry Free Response Questions**

Exam Information (AP Chemistry Exam from spring 2013 and prior) Exam Content The AP Chemistry Exam covers a full-year introductory college course in chemistry with laboratory. The multiple-choice questions in Section I cover the breadth of the curriculum. Section II, the free-response part of the exam, includes three quantitative problems, one question on writing chemical reactions and ...

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...

AP\* Stoichiometry Free Response Questions page 2 1993 I.  $2 \text{Mn}^{2+} + 4 \text{OH}^- + \text{O}_2(\text{g}) \rightarrow 2 \text{MnO}_2(\text{s}) + 2 \text{H}_2\text{O}$  II.  $\text{MnO}_2(\text{s}) + 2 \text{I}^- + 4 \text{H}^+ \rightarrow \text{Mn}^{2+} + \text{I}_2(\text{aq}) + 2 \text{H}_2\text{O}$  III.  $2 \text{S}_2\text{O}_3^{2-} + \text{I}_2(\text{aq}) \rightarrow \text{S}_4\text{O}_6^{2-} + 2 \text{I}^-$  The amount of oxygen,  $\text{O}_2$ , dissolved in water can be determined by titration. First,  $\text{MnSO}_4$  and  $\text{NaOH}$  are added to a sample of water to convert all of the dissolved  $\text{O}_2$  to  $\text{MnO}_2$ , as shown in ...

## **AP\* Stoichiometry Free Response Questions**

2020 Free-Response Questions. 2020 free-response questions are now in the AP Classroom question bank for teachers to assign to students as homework or in class, and do not require secure assessment. These questions have been updated, where possible, to best match the format of free-response questions in the course and exam description and on traditional AP Exams.

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