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Online Test of Chapter 5 Periodic Classification of Elements MCQ Science| Class 10th Question :- 1. How many elements are present in the periodic table? (a) 101 (b) 112 (c) 117 (d) 118 2. Newland's rule known as \_\_\_\_ (a) Law of triads (b) Law of octaves (c) Law of periodic table (d) None of these...

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### **Chapter 5 Test The Periodic Table Multiple Choice Answer ...**

name \_\_\_\_ HPS \_\_\_\_ # \_\_\_\_ date: \_\_\_\_ Chapter 5 TEST: The Periodic Table Multiple Choice Identify the choice that best completes the statement or answers the question. \_\_\_\_ 1. The order of elements in the periodic table is based on a. the number of protons in the nucleus. b. the electric charge of the nucleus. c. the number of neutrons in the ...

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

### 5 The Periodic Law

Holt McDougal Modern Chemistry Chapter Test Assessment Chapter Test B Teacher Notes and Answers 5 The Periodic Law TEST B 1. a 2. c 3. d 4. d 5. a 6. a 7. c 8. a 9. lanthanides 10. 2 11. fourth 12. transition elements 13. 32 14. valence electrons 15. electron affinity 16. electronegativity 17. ionization energy 18. 3 s<sup>2</sup> 3p<sup>4</sup> 19. atomic radius 20 ...

### Assessment Chapter Test B

Chapter Test A Chapter: The Periodic Law Use the periodic table below to answer the questions in this Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_\_ 1. Mendeleev organized the chemical elements based on their a.

### Assessment Chapter Test A

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5 Chapter Test (continued) , Questions For each of the following pairs, circle the atom or ion that

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has the larger atomic radius. 17.  $\text{Na}^+$  18.  $\text{NO}^-$  19.  $\text{O}^{2-}$  20.  $\text{O}^-$  Answer each of the following 4 questions as directed. 21. An element has an electron configuration of  $[\text{Ar}]4s$ . 2. Without consulting the periodic table, determine this element's group ...

### Name Date Class ~. 5 Chapter Test'

\_\_\_\_ 5. Mendeleev is credited with developing the first successful. a. periodic table. b. method for determining atomic number. c. test for radioactivity. d. use of X rays. \_\_\_\_ 6. Mendeleev did not always list elements in his periodic table in order of increasing atomic mass because he grouped together elements with similar. a. properties. c ...

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Online Test of Chapter 5 Periodic Classification of Elements MCQ Science| Class 10th Question :- 1. How many horizontal rows or periods are present in the modern periodic table? (a) 5 (b) 7 (c) 6 (d) 8 2. Maximum number of electron filled in a shell is given by \_\_\_\_ (a)  $2n^2$  (b)  $3n^2$  (c)...

### Chapter - 5 Periodic Classification of Elements MCQ Test 2 ...

Chapter 5 TEST: The Periodic Table Multiple Choice Identify the choice that best completes the statement or answers the question. \_\_\_\_ 1. The order of elements in the periodic table is based on a. the number of protons in the nucleus. b. the electric charge of the nucleus. c. the number of neutrons in the nucleus. d. atomic mass. \_\_\_\_ 2.

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### **Chapter 5 Review The Periodic Law Section 3**

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### **Holt McDougal Modern Chemistry Chapter 5: The Periodic Law ...**

Chapter 5 - The Periodic Law 5-1 History of the Periodic Table I. Mendeleev's Periodic Table A. Organization 1. Vertical columns in atomic weight order a. Mendeleev made some exceptions to place elements in rows with similar properties (1) Tellurium and iodine's places were switched 2. Horizontal rows have similar chemical properties

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