

## Chemactivity 9 Answers

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9.Because every H atom and ion must have one proton, ... ChemActivity 9. 1.a) No. b) There are three peaks. Two peaks have a low IE with a 3:1 ... answer depends on the students' answer to CTQ 3. 7.The diagram should show 3 spin-up electrons at the 2. p. level. N must have more unpaired electrons than C. 8.They tend not to pair. 9.a ...

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4.The bond energy (in kJ/mole) is the energy required to break one mole of the specified bonds. Triple bonds share six electrons between two atoms and are stronger (have greater bond energy) than double bonds, which share four electrons between two atoms.

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Read PDF Chemactivity 9 Shell Model Answers research. as [Books] Chemactivity 9 Shell Model Answers ChemActivity 9 1. a) Two. b) Lower energy peak (1s) is 2 x the intensity of the higher energy peak (2s). c) The nuclear charge for H, He, and Li is 1, 2, and 3, respectively. Therefore, the electrons in the first shell will be held most tightly ...

**Chemactivity 9 Shell Model Answers**  
ChemActivity 4 1. No. The ionization energy of He would be about 4x (twice the charge and half the distance) the ionization energy of H if this were the case. 2. Several possible answers. A typical one would have all three electrons at a much farther distance than in H from the nucleus – at least six or seven times as far. Problem 1. a) V =

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ChemActivity 3 Coulombic Potential Energy 9 5. Recall that a 1H atom consists of a proton as the nucleus and an electron outside of the nucleus. Is the potential energy, V, of a hydrogen atom a positive or negative number? Explain your answer. Model 2: Ionization Energy.

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ChemActivity 28 The Mole concept 7. How many carbon atoms are found in one mole of methane molecules? C 23 8. How many hydrogen atoms are found in one mole of methane molecules? 167 9. Calculate the average mass (in am ofone methane molecule (to 0.01 amu). 10. Based on your answer to CTQ 9, calculate the mass (in grams) of one mole of

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ChemActivity 29 Chemical Equations Exercises 175 6. How many grams of Cr2S3 are produced when the reaction in Ex. 1a (above) occurs with 10.0 grams 7. How many grams of hydrogen s lide are product (v \*6365 grams of carbon disulfide are consumed in tbf regio tin E. b. m of hydrogen 8. How many grams f Iron c or e are pro u FeC)3

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ChemActivity 14 Bond Order and Bond Strength 9. Rank N2, P2, As2, in order from weakest to strongest bond.10. Indicate whether each of the following statements is true or false and explain your reasoning.

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Based on your answer to CTQ 6, what do all nickel (Ni) atoms have in common? ChemActivity 1 The Nuclear Atom 3. 8. What structural feature is different in isotopes of a particular element? 9. How is the mass number, A, (left-hand superscript next to the atomic symbol as shown in ...

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Answers to Critical Thinking Questions Please do not give these answers to students. Quite often, the answers given here are less detailed than what is expected for a student's answer. ChemActivity 1 1. 6, 6, 6, 2, 6, 7, 3, 6, 6, 7, 4. All carbon atoms and ions have six protons in the nucleus. 5.

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