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course, the student should be able to:
Describe laboratory procedures and give the names and uses of laboratory equipment to allow for the

**CHEMISTRY 1120 -
GENERAL
CHEMISTRY I**

Corequisite: CHEM 1120. (This class can be taken prior to CHEM 1121 with a grade of at least C-). Course

Description:

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Continuation of CHEM 1111. Experiments and experimental techniques in general chemistry. Textbook and other materials: CHEM 1121 Laboratory Manual

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The reagent quantities tabulated in the lab manual for 0.20 M I^- , $0.10 \text{ M S}_2\text{O}_8^{2-}$, DI Water 3% starch and $0.0050 \text{ M S}_2\text{O}_3^{2-}$ were combined according to the lab manual produced in a 250mL flask. The reaction was monitored by recording the time in seconds required for the solution to change from colorless to purple for each respective trial.

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Introduction to General
Chemistry I Laboratory
General Chemistry I

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Laboratory (CHM151L) is designed to be taken by students enrolled in General Chemistry I Lecture (CHM151). The laboratory and lecture are separate courses and you will be assigned a separate grade for each. CHM151L will often reinforce lecture course topics,

**Introduction to
General Chemistry I
Laboratory**

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After completing the requirements of CHEM 1120, students will be able to:

- Conduct an experiment, collect and analyze data, and interpret results in a laboratory setting.
- Analyze, test, and evaluate a scientific hypothesis.

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**CHEMISTRY II Term
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Syllabus - CHEM 1120

Laboratory General

Chemistry II Spring

Semester - 2013

Instructor: Dr. James B.

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the laboratory manual.

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completes the

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presented to the

student for completion.

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chemical exposure in
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Grading Grades will be
determined by
performance on three
1.5 -hour exams, a
2-hour final exam,
Case Study/Lab
assignments, e-

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homework, and participation in discussion sections. Plus and Minus grades will be assigned. The College and EPC recommends that the A+ grade will be used only to

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Introductory Chemistry
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Lab reports are due the

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week of November 14 - 18, 2016 Your lab report must contain the following information: Experimental data may be collected with other students in introductory chemistry labs. However, I understand that sharing information required for a lab report or unknown submission (including but not limited to word processing or ...

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CHEMISTRY 1120
(4:012) - PRINCIPLES
OF CHEMISTRY II -
Spring 2016 . Lectures
A & B, and Lecture C
(Honors) Professors
Nicole Becker, Amy
Strathman, Alexandra
Brandriet and Jordan
Harshman . Lecture A:
8:30 - 9:20 AM, MWF in
W10 PBB Lecture B:
11:30 AM -12:20 PM,

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MWF in W290 CB

Lecture C (Honors):

8:30 – 9:20 AM, MWF in
W151 PBB . Professor

**CHEMISTRY 1120 -
PRINCIPLES OF
CHEMISTRY II - FALL
2015**

Name_ CHEMISTRY

1120 LAB PROBLEM

SET-REDOX

EQUATIONS I. BALANCE

by CHANGE in

OXIDATION NUMBER +

H₂O ZnS + HNO₃ ?

Zn(NO₃)₂ + S + NO +

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H₂O . Get more help from Chegg. Get 1:1 help now from expert Chemistry tutors ...

Solved: Name_
CHEMISTRY 1120
LAB PROBLEM SET-
REDOX EQUATIO ...

Place 3ml of Na₂CO₃ in to a small test tube and place a rubber stopper and glass tube into the open end.

Place 10ml in Ca (OH)₂ into a large test tube.

next put 6 drops of HCl

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in to the tube w/
 Na_2CO_3 and quickly
place the rubber
stopper i=and tube
back in. then place the
other end of the hollow
tube in to the
limewater and observe.

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