

Chemquest 39 Answer Key

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Prove that $X_{\text{solute}} + X_{\text{solvent}} = 1.2$ The answer from question 12 + answer to question 13 = 115. In a certain salt water solution, the mole fraction of salt is 0.18. Find the mole fraction of water. $1.00 - 0.18 = 0.82$ Information: Mass Percent CompositionMass percent composition is similar to the mole fraction except the amounts of solute and solvent are in grams instead of moles.

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Therefore, the charge on each X is 1. Here's why: Remember Answer: + that S has a negative 2 charge. There are two "X" ions to balance out the negative 2 charge of S.

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