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ANSWER: Correct Chapter 9 Reading Question 2 Part A Iron reacts with oxygen in the air to produce rust ( $\text{Fe}_2\text{O}_3$ ) according to the reaction shown below.  $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$  How many moles of rust can be formed if 1.8 mols of iron are present? Hint 1. Begin with your given number of moles.

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