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water, forming solid titanium(IV) oxide and aqueous hydrochloric acid. 13.

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Modeling Chemistry Ws Answers Unit 9

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1. What volume does 16.0 g of O₂ occupy at STP? 2. A mixture contains 5.00 g each of O₂, N₂, CO₂, and Ne gas. Calculate the volume of this mixture at STP 3.

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Page 2 of 3. Modeling Chemistry 2 U8 ws 3 v2.0 6. How many grams of silver nitrate (you need the formula to get the molar mass) are needed to prepare 250 mL of standard 0.100 M silver nitrate solution? (4.25g) 7. If 10.0 g of AgNO₃ is available, find the volume needed to prepare a 0.25 M AgNO₃ solution. (0.235 L) 8.

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Modeling Chemistry U9 Ws 2 V2 Answers - Company Chemistry I Name ____ Unit 1 WS 1 - Measurement and Significant Figures For each of the following, write the appropriate measurement,

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then the number of significant figures in the reading.

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Modeling Chemistry 1 U9 WS1 2013 Name Date Pd Chemistry - Unit 9 Worksheet 1 - Gases Again Use gas laws to solve # 1 - 3, doing the work like we did in Unit 2 (PVT chart). Use new information to solve # 4 - 8. 1. A can of spray paint contains nitrogen gas as the propellant. The pressure of the gas is 3.5 atm

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Modeling Chemistry 2 U9 ws1 v2.0 Elemental metals exist as single, unbonded atoms. (Ex: formula for copper metal is Cu) Watch for ionic vs molecular compounds. Use nomenclature rules, and your ion chart and periodic table to figure out the formulas for these.

General Chemistry - Unit 9 Worksheet 1

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