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Review Purposes. To get an overall view of stoichiometry. Apply skills learned to perform quantitative chemical analysis. Apply theories and rules of chemistry to solve problems. Assess areas of strength and weakness for review purposes. Improve problem solving strategy and learning efficiency.

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4 a. What is the value of the coefficient x in this equation? 40.07 g/mol b. What is the molar mass of C_3H_4 ? 2 mol O₂:1 mol H₂O c. What is the mole ratio ...

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Date Chapter Test B, continued 31. Calculate the energy released by freezing 13.8 g of a liquid. Its molar mass is 82.9 g/mol, and its molar enthalpy of fusion is 4.60 kJ/mol. PART VI Define the term equilibrium vapor pressure.

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