

Ocr F325 June 2010 Question Paper

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Ocr F325 June 2010 Question

F325 Mark Scheme June 2010 6 Question Expected Answers Marks Additional Guidance d $\text{HNO}_3 + \text{CH}_3\text{CH}_2\text{COOH} \rightleftharpoons \text{CH}_3\text{CH}_2\text{COOH}_2 + + \text{NO}_3^-$ acid 1 base 2 acid 2 base 1 2 State symbols NOT required ALLOW 1 AND 2 labels the other way around. ALLOW 'just acid' and 'base' labels throughout if linked by lines so that

Mark Scheme for June 2010 - AQA, OCR and Edexcel

† Write your answer to each question in the space provided. Additional paper may be used if necessary but you must clearly show your Candidate Number, Centre Number and question number(s). INFORMATION FOR CANDIDATES † The number of marks is given in brackets [] at the end of each question or part question.

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F325 Equilibria Energetics and Elements June 2010 Q7

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F325 June 2010

F325 Mark Scheme June 2010 7 Question Expected Answers Marks Additional Guidance 4 a i Complete circuit (with voltmeter) and salt bridge linking two half-cells □ Pt electrode in solution of $\text{Fe}^{2+}/\text{Fe}^{3+}$ □ Ag in solution of Ag^+ □ 3 DO NOT ALLOW 'solution of a silver halide', e.g. AgCl (as these are insoluble) but

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F325 Mark Scheme June 2011 1 Question Answer Mark Guidance 1 (a) (The enthalpy change that accompanies) the formation of one mole of a(n ionic) compound from its gaseous ions (under standard conditions) 2 IGNORE 'Energy needed' OR 'energy required' ALLOW as alternative for compound: lattice, crystal, substance, solid Note: 1st mark requires 1 mole

F325 Mark Scheme June 2011 Question Answer Mark Guidance 1 ...

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Advanced GCE Unit F325: Equilibria, Energetics and Elements

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F325 Mark Scheme January 2012 6 Question Expected answers Marks Additional guidance 2 a (The enthalpy change that accompanies) the formation of one mole of a(n ionic) compound from its gaseous ions (under standard conditions) 2 IGNORE 'Energy needed' OR 'energy required' ALLOW as alternative for compound: lattice, crystal, substance, solid, product

Advanced GCE Unit F325: Equilibria, Energetics and Elements

F325 Mark Scheme June 2010 8 Question Expected Answers Marks Additional Guidance c A fuel cell converts energy from reaction of a fuel with oxygen into a voltage/electrical energy □ 5 ANNOTATIONS MUST BE USED ALLOW combustion for reaction of fuel with oxygen/reactants ALLOW a fuel cell requires constant supply of fuel

3 ALLOW F 1450 B 736 G E 76 D -642 FIVE FOUR THREE b ALLOW ...

F325 Mark Scheme June 2011 1 Question Answer Mark Guidance 1 (a) (The enthalpy change that accompanies) the formation of one mole of a(n ionic) compound from its gaseous ions (under standard conditions) 2 IGNORE 'Energy needed' OR 'energy required' ALLOW as alternative for compound: lattice, crystal, substance, solid Note: 1st mark requires 1 mole

Advanced GCE Unit F325: Equilibria, Energetics and Elements

F325 Mark Scheme June 2014 3 Question Answer Marks Guidance 1 (a) (i) $2\text{K}^+(\text{g}) + \text{S}^-(\text{g}) + \text{e}^- \rightleftharpoons 2\text{K}^+(\text{g}) + \text{S}^{2-}(\text{g})$ $2\text{K}(\text{g}) + \text{S}(\text{g})$ 3 Mark each marking point independently Correct species AND state symbols required for each mark For S^{2-} , DO NOT ALLOW $\text{S}^{=2}$ For e^- , ALLOW e For e^- only, IGNORE any state symbols added ALLOW k and s It can be very difficult distinguishing K from k; S from s

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