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this set (27) A chemical

compound always has

the same elements in

the same proportions

by mass regardless of

the source of the

compound. This is a

statement of _____

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_____ Section _____

Date _____ 10. A

student drew the
following electron box

...

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Date _____ 1. Which of the following statements is true? A. An equation is balanced by changing the subscripts as needed. B. Matter is usually, but not always, conserved in a balanced chemical equation.

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The test data is plotted
in the graph below. 0

20 40 60 80 100 120

140 160 0 1020 3040

50 60 70 80 90

Solubility (g/100 g H₂O)

Temperature (°C)

Solubility of Various

Solids Versus

Temperature KNO₃

NaNO₃ NH₄Cl NaCl

Na₂SO₄ KNO₃ NaNO₃

NH₄Cl Na₂SO₄ NaCl

Key

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compounds that
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practice, use the
following passage and
graphs to answer
questions 59 a student
conducted a series of
experiments to
determine how the

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solubility of several solid substances varied with temperature she tested.

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question is indicated
by a . 1: A solid that
forms from a solution
during a chemical
reaction is called a
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provide a useful study tool for students. For instructors, they are a good resource for homework, quiz, or test questions Standardized test prep chemistry chapter 4 answers.

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Practice Test Multiple
Choice (3 pts each): _

1. Which of the
following is NOT a
chemical

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ecf8427e.