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definite shape & definite volume. Liquid.  
state of matter in which materials have  
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Law. energy. kinetic energy. potential energy. thermal energy

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A state of matter that has a definite volume but not a definite shape. A state of matter that does not have a definite volume or shape. A state of matter that

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has both a fixed volume and shape.

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states of matter: solid, liquid, and gas. Examples of solids include a table, a pencil, and ice.

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contrast solids, liquids, and gases based on the basic properties of each of these states of matter.

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4. What causes matter to change states?  
The temperature determines the state of a substance. When heated enough, a

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solid changes to a liquid, and a liquid to a gas. These changes of state take places because the molecules, or tiny bits of matter than make up the substance, vibrate faster when the substance is heated (kinetic energy increases).

## **Properties of Matter Questions |**



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## **Shmoop**

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DRAFT. 6th grade. 749 times. Religious Studies. 69% average accuracy. a year ago. lindaestes41. 0. Save. Edit. Edit. ...  
Different states of matter have different amounts of space between the atoms.  
Which list ranks the states of matter, ...

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MATTER TEST IS ON \_\_\_\_\_ \*\*HEAT-see  
heat notes along with study guide.\*Know  
the 3 ways to produce heat: rubbing,  
burning, and using electricity RUBBING:  
when objects are rubbed together, heat  
is produced. Ex: when hands are rubbed

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together, they get warmer on a cold day.

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Atmospheric pressure Pressure exerted  
by mercury column 760 mm Name Date  
Class Study Guide for Content Mastery

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Chemistry: Matter and Change Chapter  
13 73 States of Matter Section 13.1  
Gases In your textbook, read about the  
kinetic-molecular theory. Complete each  
statement. 1.

**13 STUDY GUIDE FOR CONTENT  
MASTERY 13 STUDY GUIDE FOR ...**  
Physical, Chemical, and States of Matter

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Study Guide. Flashcard maker : James Storer. Physical Change. changes the shape or form of the object or material without changing it into a new substance. Chemical change. changes the identity of one or more substances so that a new substance is formed.

**Physical, Chemical, and States of**

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CHAPTER 10 REVIEW States of Matter  
SECTION 3 SHORT ANSWER Answer the following questions in the space

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provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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## **10 States of Matter - Ms. Agostine's Chemistry Page**

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## **States Of Matter Study Guide Answer Key**

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